

1. Introduction

The time constant T₂ is the transverse or spin-spin relaxation time. Any process that causes loss of magnetization on the x-y plane contributes to T₂. The magnetization follows an exponential decay:

$$M = M_0 e^{-t/T_2} \quad (1)$$

However, the inhomogeneity of the static field also causes a loss of the transverse magnetization, which is written as T₂^{*}. Since spin echoes remove the effects of inhomogeneity, the time constant that results from the decay envelope of a series of spin echoes is the real T₂. This experiment is regarded as Carr-Purcell-Meiboom-Gill (CPMG) experiment.

(Method 1)

2. Pulse sequence

Figure 1 shows an approach to apply the CPMG method for measuring T₂: A series of echoes follows the 90° pulse and the NMR signal is acquired at the center of each echo. One scan will have *n* acquisitions for *n* echoes.

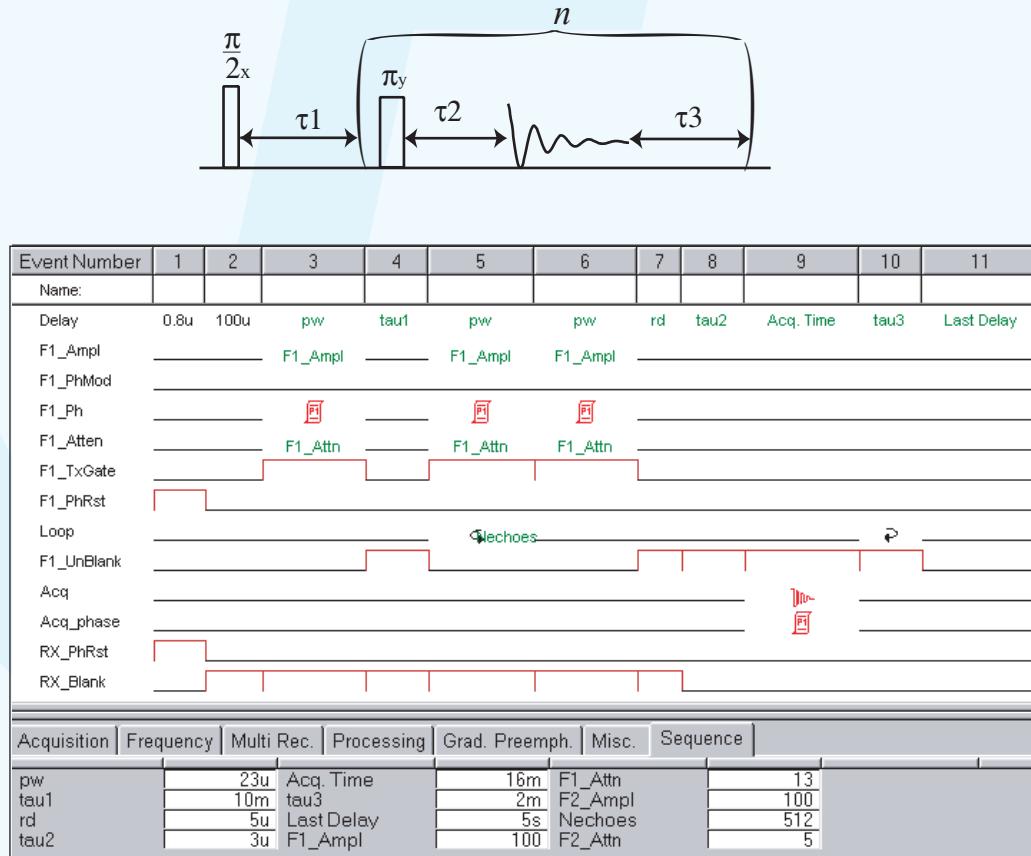


Fig. 1 The CPMG pulse sequence for Method 1.

3. Experiment and results

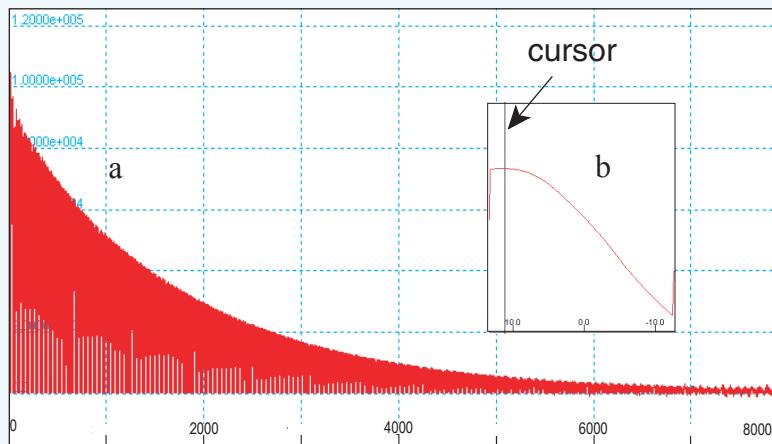


Fig. 2 a: H₂O ¹H CPMG data acquired with the pulse sequence in Fig.1. b: Expansion of one echo from a.

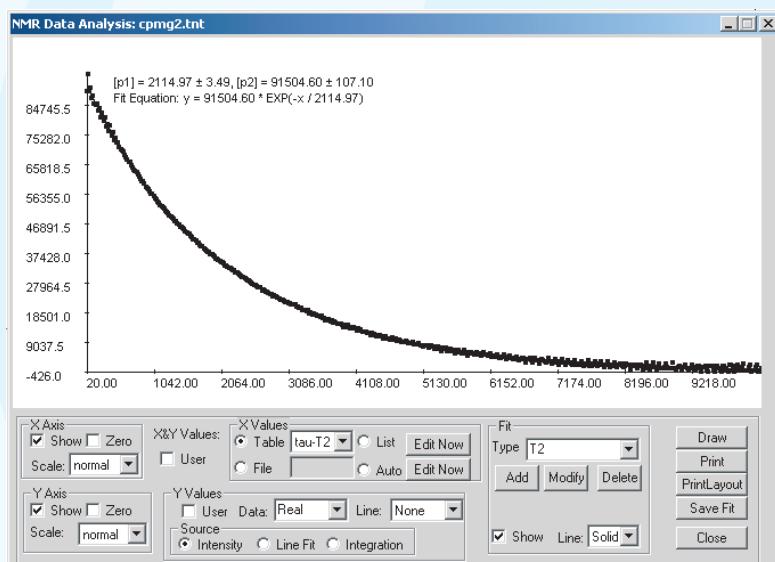


Fig. 4 NMR data Analysis window showing the T₂ fitting for H₂O and T₂ = 2.1 s.

Sample: H₂O

90° pulse = 23 μ s

τ_1 = 10 ms

τ_2 = 2 ms

τ_3 = 3 ms

Number of echoes/acquisitions: 512

Data processing:

1. Separate the data into a 512-record 2D data file using "Separate Records" (Fig.2b).
2. Create variable "Echo Interval" with a value of one echo time, and an empty 2D delay table "tau-T2" in the sequence. "Echo Interval" is introduced as a local variable with the mathematical expression, Echo Interval = 2pw + τ_2 + τ_3 .
3. Run the script "CPMG proc setup" to fill table "tau-T2" with delays for all the echoes.
4. Locate the cursor at the position of the desired intensity used to calculate T₂ at any of the records (see Fig. 2b). (As H₂O has only one peak, the FID is directly used to calculate T₂.)
5. Click on "NMR Data Analysis", select the appropriate settings, and click on "Draw".

4. Reference

1. Derome, A., "Modern NMR Techniques for Chemistry Research", Pergamon Press, New York, 1987.